	METEOROLOGY NAME	
IIN	INSTRUCTOR DATE / / LAB: ADIABATIC TEMPERATURE CHANGE	
	<b>Materials:</b> Fizz-Keeper, 1-L plastic soft-drink bottle, thermometer that fits in the bott thermometers are easy to read), water, match, (optional: flexible wire and tape).	le (liquid crystal
1.	<ul> <li>Place the thermometer in the dry bottle.</li> <li>When performing an activity that uses temperature as a variable, why should the bot little as possible?</li> </ul>	
2.	<ul> <li>2. Screw the Fizz-Keeper tightly onto the mouth of the bottle. Record the temperature inside</li> <li>Starting temperature</li></ul>	the bottle.
3.	<ul> <li>Pump the Fizz-Keeper 60 times. Record the temperature after pumping.</li> <li>Temperature after pumping</li> <li>Did the temperature increase or decrease? Why do you believe this happened?</li> </ul>	
4.	<ul> <li>4. Pump the Fizz-Keeper an additional 20 times and record any further changes in temperature</li> <li>Temperature after pumping</li> </ul>	
5.	<ul> <li>Unscrew the Fizz-Keeper. After several minutes record the temperature of the air inside the</li> <li>Temperature after removing Fizz-Keeper</li> <li>Did the temperature increase or decrease? Why do you believe this happened?</li> </ul>	
6.	<ul> <li>6. Remove the thermometer and pour a small amount of water into the bottle. Swirl the water bottle for 20 seconds, then pour out the water.</li> <li>What gases are probably present in the bottle?</li> </ul>	around inside the
7.	<ul> <li>7. Screw the Fizz-Keeper tightly onto the mouth of the bottle and pump the Fizz-Keeper 60 tremove the Fizz-Keeper.</li> <li>Is anything visible in the "space" of the bottle? If so, what?</li> </ul>	
8.	<ul> <li>8. Light a match. After it burns briefly, blow out the flame, quickly drop the smoking match and screw the Fizz-Keeper tightly onto the bottle.</li> <li>Can any particles be seen in the "space" of the bottle?</li> </ul>	
9.	<ul><li>9. Pump the Fizz-Keeper 60 times then quickly release the pressure. Look carefully for any ch</li><li>• What is visible in the "space" of the bottle?</li></ul>	_
	• Summarize the conditions (pressure and temperature changes) and materials needed in "cloud."	order to create a

How can this happen naturally in the atmosphere?